Amendments to the Specification

Please add the following new paragraph after the title and before the first paragraph on page 1 as follows:

This application is a U.S. National Phase application of PCT International Application PCT/JP2005/000630

Please replace the paragraph beginning at page 11, line 17, with the following rewritten paragraph:

The EDLCs of the present embodiment and the conventional EDLCs are also subjected to CV measurement to examine the reaction potential of the collecting electrodes containing the Al_4C_3 alloy layer of the present embodiment. The results are shown in Fig. 3. The reference electrode is an Ag/Ag+ electrode and the counter electrode is Pt. As the working electrodes, a collecting electrode containing the Al_4C_3 alloy layer, an aluminum electrode having an aluminum electrode and carbon particles fixed thereon, and an aluminum electrode are used for comparison. The results show that the collecting electrode containing the Al_4C_3 alloy layer has a more noble reaction potential than the aluminum electrode having an aluminum electrode and carbon particles fixed thereon and the other aluminum electrode. In other words, the use of the electrodes containing the Al_4C_3 alloy layer as the collectors can make the potential window larger than in the conventional electrodes. This seems to indicate that an EDLC can have a high withstand voltage by using the collecting electrodes containing the Al_4C_3 alloy layer.

Please replace the paragraph beginning at page 17, line 9, with the following rewritten paragraph:

Fig. 9 is a withstand voltage characteristic of the anode of the EDLC of the present embodiment. Here, the potential at a current value of 0.01 mA is defined as the withstand voltage of the anode. As electrolytic solution 16, TEAFB4 TEABF4 is used.

Please replace the paragraph beginning at page 22, line 8, with the following rewritten paragraph:

The EDLC shown in Fig. 12B is manufactured by vacuum-impregnating conventional element 32 with an electrolytic solution consisting, for example, of TEAFB4 TEABF4 in PC solution; inserting it into processed case 31 having aluminum fluoride layer 31a shown in Fig. 11B; and sealing it with sealing rubber 33. This EDLC is referred to as Structure 1.

Please replace the paragraph beginning at page 25, line 21, with the following rewritten paragraph:

Pressure control valve 47 is inserted in such a manner that inlet 46 for injecting the electrolytic solution is sealed after the injection. Although it is not illustrated, an electrolytic solution consisting, for example, of TEAFB4 TEABF4 in PC solution is used as the electrolytic solution.

Please replace the paragraph beginning at page 27, line 19, with the following rewritten paragraph:

Fig. 16 is a sectional view showing a method for producing a case used in an EDLC of the present embodiment. As shown in Fig. 16, aluminum case 48 is filled with fluorine-containing solution 49. Fluorine-containing solution 49 used in the present embodiment consists of TEAFB4 TEABF4 in PC solution. Fluorine-containing solution 49 has counter electrode 50 inside, which is preferably made of an electrochemically stable metal such as platinum. DC power source 51 is provided to apply a current between case 48 and counter electrode 50, and it is preferable that case 48 and counter electrode 50 have a potential difference of 3 to 5V.